

S. Thamocharan,^a
V. Parthasarathi,^{b*}
G. Anandha Babu,^b
Raveendra K. Hunnur,^c
Bharati Badami^c and
Anthony Linden^d

^aMolecular Biophysics Unit, Indian Institute of Science, Bangalore 560 012, India, ^bSchool of Physics, Bharathidasan University, Tiruchirappalli 620 024, India, ^cPost-Graduate Department of Studies in Chemistry, Karnatak University, Dharwad 580 003, India, and ^dInstitute of Organic Chemistry, University of Zürich, Winterthurerstrasse 190, CH-8057 Zürich, Switzerland

Correspondence e-mail: vpsarati@yahoo.com

Key indicators

Single-crystal X-ray study
T = 160 K
Mean $\sigma(\text{C}-\text{C}) = 0.006 \text{ \AA}$
R factor = 0.047
wR factor = 0.105
Data-to-parameter ratio = 11.9

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

5-Phenoxymethyl-1,3,4-oxadiazole-2(3H)-thione

In the title compound, $\text{C}_9\text{H}_8\text{N}_2\text{O}_2\text{S}$, the H atom of the thiol group has been transferred to the neighbouring N atom of the oxadiazole ring. Intermolecular $\text{N}-\text{H}\cdots\text{N}$ hydrogen bonds exist between adjacent molecules.

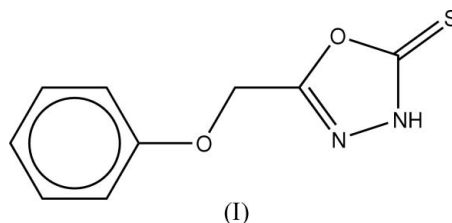
Received 10 October 2005

Accepted 14 October 2005

Online 19 October 2005

Comment

It is well known that 1,3,4-oxadiazole-2-thione derivatives show a broad spectrum of biological activities (Ram & Vlietinck, 1988; Boschelli *et al.*, 1993). A view of the title compound, (I), with the atomic numbering scheme, is shown in Fig. 1. The dihedral angle between the mean planes of the benzene and 1,3,4-oxadiazole rings is $14.4(1)^\circ$. In (I), the bond lengths and angles are in good agreement with the expected values (Allen *et al.*, 1987). The H atom of the thiol group has been transferred to the adjacent N atom of the oxadiazole ring. The $\text{N}3-\text{N}4$ [$1.383(5) \text{ \AA}$] and $\text{C}2=\text{S}2$ [$1.647(4) \text{ \AA}$] bond lengths correspond to the usual single $\text{N}-\text{N}$ and double $\text{C}=\text{S}$ distances.



The crystal structure of (I) is shown in Fig. 2. In the solid state, atom N3 is involved in an intermolecular $\text{N}-\text{H}\cdots\text{N}$ hydrogen bond with atom N4 of the oxadiazole group of an adjacent molecule (Table 1). This hydrogen bond links the molecules into chains, which run parallel to the [010] direction and can be described by a $C(3)$ graph-set motif (Bernstein *et al.*, 1995). Atom C6 (*via* H61) acts as a donor for a weak

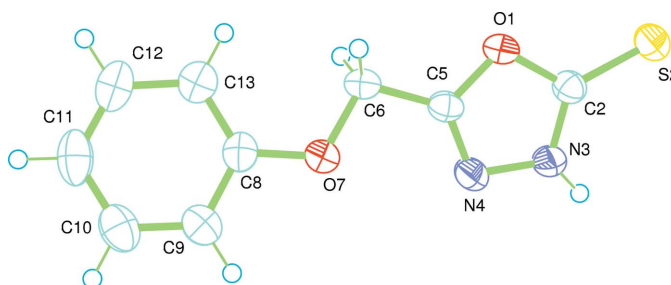


Figure 1

A view of (I), with the atom-numbering scheme. Displacement ellipsoids are drawn at the 50% probability level. H atoms are shown as small spheres of arbitrary radii.

intermolecular C—H···O interaction with atom O1 of a symmetry-related molecule. This weak interaction connects the molecules into chains, which also run parallel to the [010] direction and can be described by a graph-set motif of $C(4)$. In addition, atom C6 (*via* H62) is involved in an intermolecular C—H··· π interaction with the benzene ring of a neighbouring molecule [$H62\cdots Cg = 2.71 \text{ \AA}$, $C6\cdots Cg = 3.459(4) \text{ \AA}$ and $C6-H61\cdots Cg = 132^\circ$, where Cg is the centroid of the benzene ring at $(x, y - 1, z)$].

Experimental

A solution of phenoxyacetic acid hydrazide (0.01 mol) was dissolved in pyridine (10 ml), and carbon disulfide (5 ml) was added with constant stirring. Stirring was continued for 36 h at room temperature. The reaction mixture was then poured into ice-cold water and acidified with dilute HCl. The solid, (I), separated, was filtered off and crystallized from dimethylformamide (m.p. 457–459 K).

Crystal data

$C_9H_8N_2O_2S$	$D_x = 1.470 \text{ Mg m}^{-3}$
$M_r = 208.23$	Mo $K\alpha$ radiation
Monoclinic, $P2_1$	Cell parameters from 21914 reflections
$a = 9.3233(8) \text{ \AA}$	$\theta = 2.0\text{--}25.0^\circ$
$b = 4.9446(5) \text{ \AA}$	$\mu = 0.32 \text{ mm}^{-1}$
$c = 10.2051(10) \text{ \AA}$	$T = 160(2) \text{ K}$
$\beta = 91.388(5)^\circ$	Needle, colourless
$V = 470.32(8) \text{ \AA}^3$	$0.35 \times 0.10 \times 0.02 \text{ mm}$
$Z = 2$	

Data collection

Nonius KappaCCD area-detector diffractometer	1553 independent reflections
φ and ω scans with κ offsets	1295 reflections with $I > 2\sigma(I)$
Absorption correction: multi-scan (SORTAV; Blessing, 1995)	$R_{\text{int}} = 0.083$
$T_{\text{min}} = 0.676$, $T_{\text{max}} = 0.999$	$\theta_{\text{max}} = 25.0^\circ$
6143 measured reflections	$h = -11 \rightarrow 11$
	$k = -5 \rightarrow 5$
	$l = -11 \rightarrow 12$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0334P)^2 + 0.2737P]$
$R[F^2 > 2\sigma(F^2)] = 0.047$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.105$	$(\Delta\rho)_{\text{max}} < 0.001$
$S = 1.08$	$\Delta\rho_{\text{max}} = 0.19 \text{ e \AA}^{-3}$
1553 reflections	$\Delta\rho_{\text{min}} = -0.19 \text{ e \AA}^{-3}$
131 parameters	Absolute structure: Flack & Bernardinelli (1999, 2000), 636 Friedel pairs
H atoms treated by a mixture of independent and constrained refinement	Flack parameter: 0.01 (14)

Table 1

Hydrogen-bond geometry (\AA , $^\circ$).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$N3-H3\cdots N4^i$	0.87 (5)	2.24 (4)	2.899 (5)	132 (3)
$C6-H61\cdots O1^{ii}$	0.99	2.53	3.416 (4)	148

Symmetry codes: (i) $-x + 1, y - \frac{1}{2}, -z$; (ii) $-x, y + \frac{1}{2}, -z$.

The position of the amine H atom was determined from a difference Fourier map and refined freely along with its isotropic displacement parameter. All remaining H atoms were placed in geometrically idealized positions and were constrained to ride on

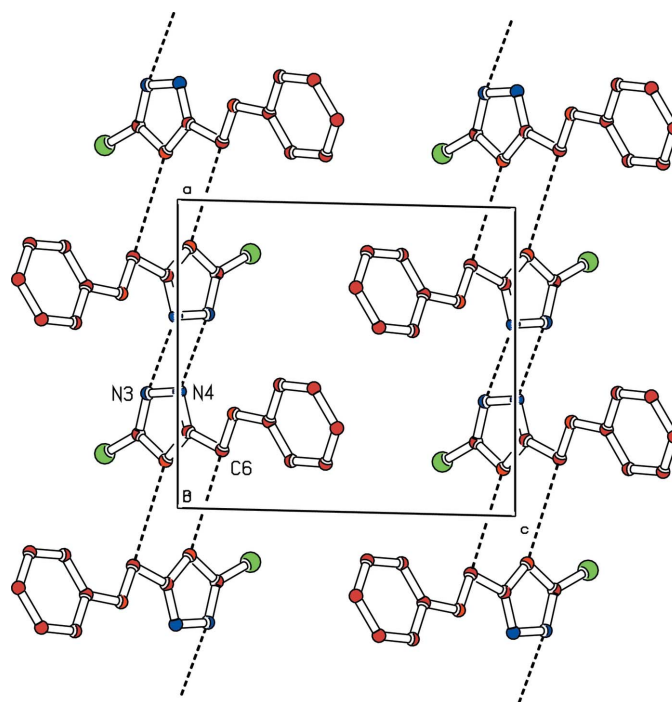


Figure 2

Crystal structure of (I), as projected on to the ac plane. N—H···N and C—H···O bonds are indicated by dashed lines. H atoms have been omitted.

their parent atoms, with C—H distances in the range 0.95–0.99 \AA and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$.

Data collection: *COLLECT* (Nonius, 2000); cell refinement: *DENZO-SMN* (Otwinowski & Minor, 1997); data reduction: *DENZO-SMN* and *SCALEPACK* (Otwinowski & Minor, 1997); program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1994); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3* (Farrugia, 1997); software used to prepare material for publication: *SHELXL97* and *PLATON* (Spek, 2003).

References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–19.
- Altomare, A., Casciarano, G., Giacovazzo, C., Guagliardi, A., Burla, M. C., Polidori, G. & Camalli, M. (1994). *J. Appl. Cryst.* **27**, 435.
- Bernstein, J., Davis, R. E., Shimoni, L. & Chang, N.-L. (1995). *Angew. Chem. Int. Ed. Engl.* **34**, 1555–1573.
- Blessing, R. H. (1995). *Acta Cryst.* **A51**, 33–38.
- Boschelli, D. H., Connor, D. T., Bornemeier, D. A., Dyer, R. D., Kennedy, J. A., Kuipers, P. J., Okonkwo, G. C., Schrier, D. J. & Wright, C. D. (1993). *J. Med. Chem.* **36**, 1802–1810.
- Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
- Flack, H. D. & Bernardinelli, G. (1999). *Acta Cryst.* **A55**, 908–915.
- Flack, H. D. & Bernardinelli, G. (2000). *J. Appl. Cryst.* **33**, 1143–1148.
- Nonius (2000). *COLLECT*. Nonius BV, Delft, The Netherlands.
- Otwinowski, Z. & Minor, W. (1997). *Methods in Enzymology*, Vol. 276, *Macromolecular Crystallography, Part A*, edited by C. W. Carter Jr & R. M. Sweet, pp. 307–326. New York: Academic Press.
- Ram, V. J. & Vlietinck, A. J. (1988). *J. Heterocycl. Chem.* **25**, 253–256.
- Sheldrick, G. M. (1997). *SHELXL97*. University of Göttingen, Germany.
- Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.